

Basic Chromatography

Basic Chromatography

- A chromatograph is a computerized machine that analyzes hydrocarbons, chemicals, and permanent gases.
- During one analysis cycle a lot of components are separated and analyzed individually
- The information from the chromatograph includes the mole percent or liquid volume percent of the components and/or parts per million of the components that make up the sample.
- The goal of an online process gas chromatograph is to supply quantitative and qualitative information.
- The information is used to not only monitor and control the process but to optimize its use of energy, time and raw materials.
- PGC (Process GC) : Online but discontinuous

Basic Chromatography

A basic Gas Chromatograph (GC) consists of the following parts:

- Carrier Gas
- Carrier Regulator
- Sample Valve
- Column
- Detector
- Sample Conditioning Systems
- System Controller

Basic Chromatography

■ Sample Conditioning Systems

- Sample probe
- Sample Transport Tubing
- Sample Phase Consistency
 - Filtration
 - Liquid Coalescing
 - Heat Tracing

■ Speed Loops

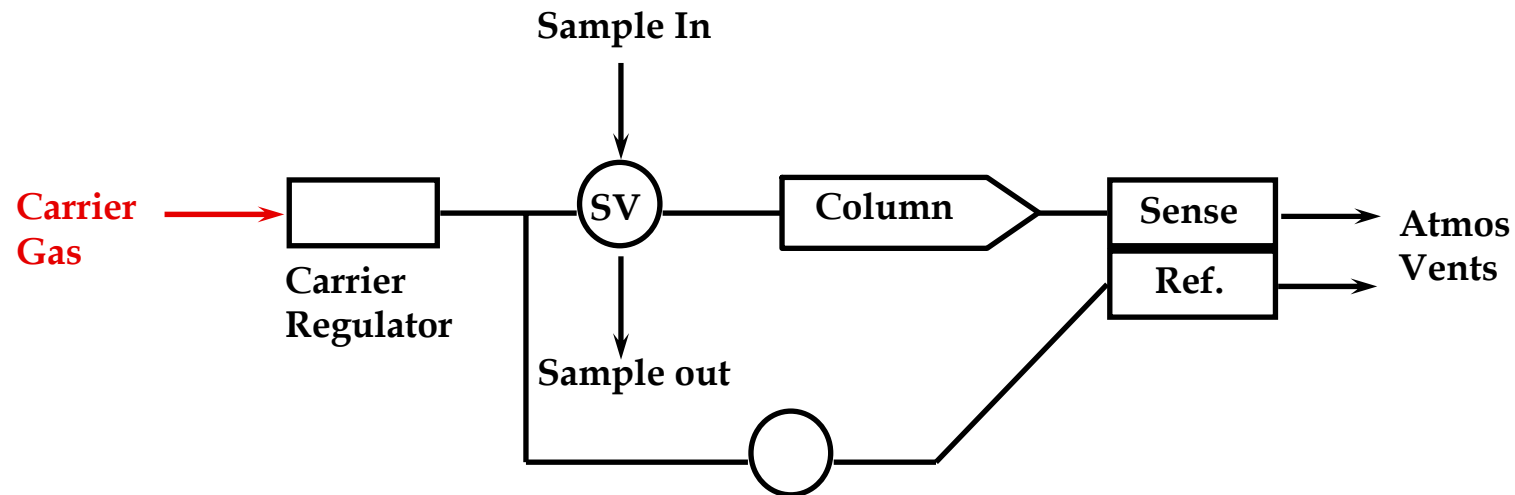
■ Automatic Stream switching

- Calibration Standards
- Multiple Sample Streams

Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Carrier Gas

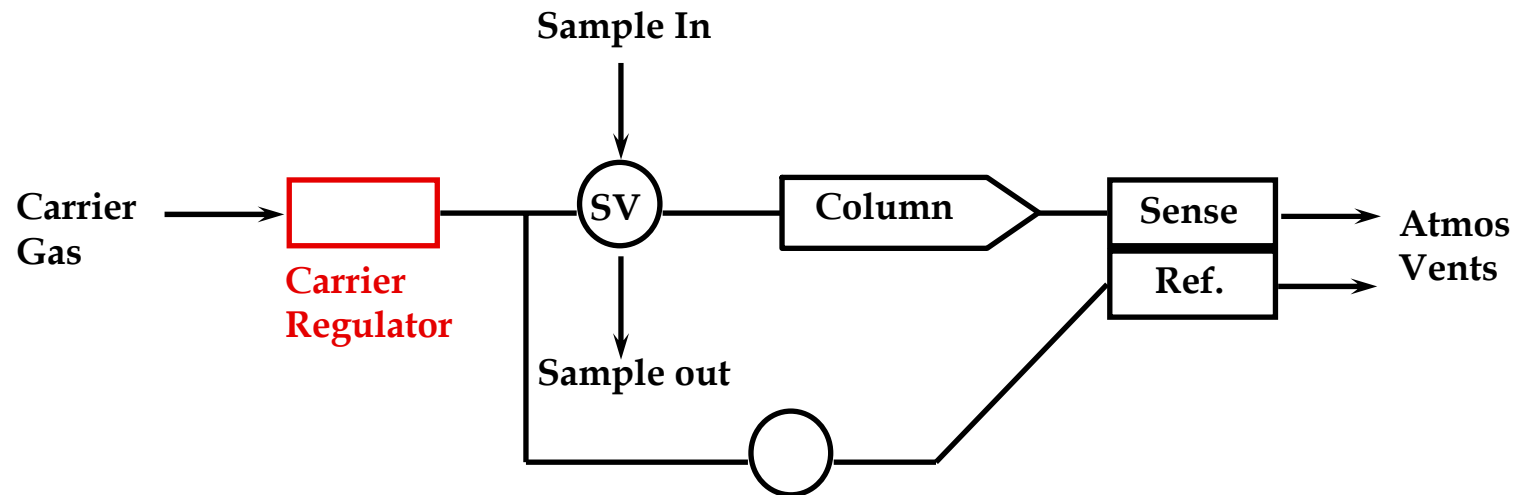
Carries the sample through the column and detector to an atmospheric vent.



Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Carrier Regulator

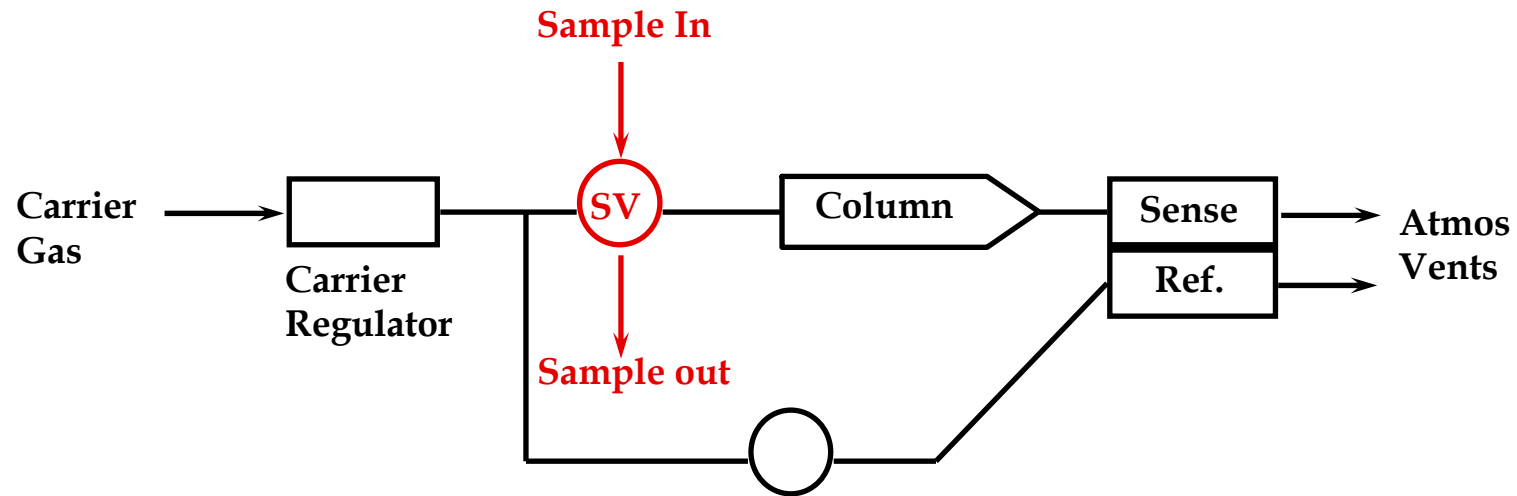
Maintains a constant pressure of carrier gas which results in a constant carrier flow rate.



Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Sample Valve

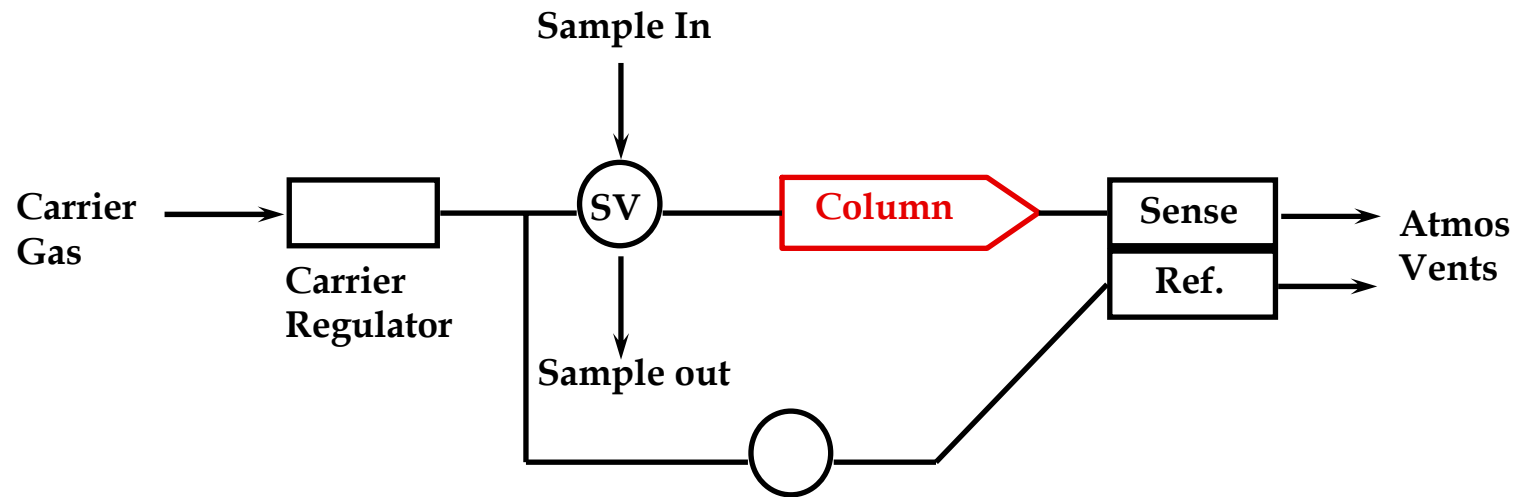
Injects a measured amount of sample.



Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Column

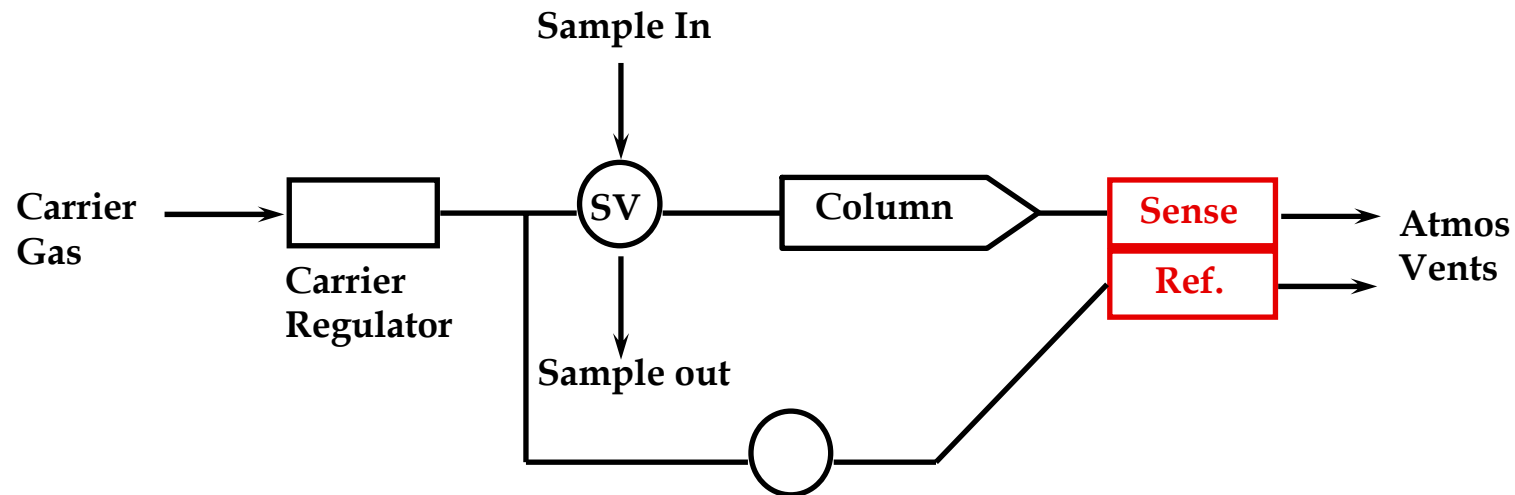
Separates the sample into individual components.



Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Detector

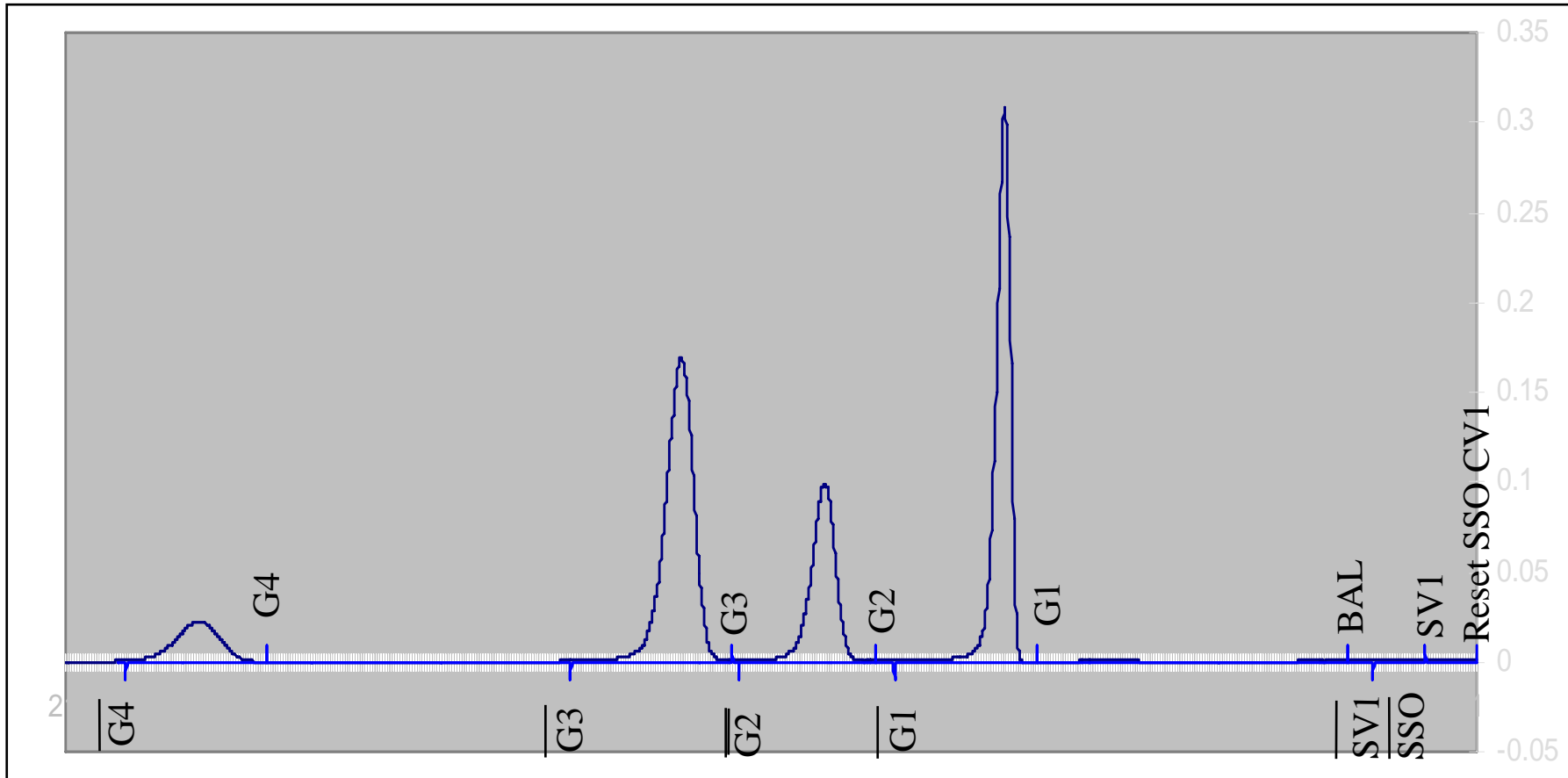
Senses the individual components as they elute off the column.



Basic Chromatography

■ Gas Chromatogram:

- The trace of the detector signal drawn on a recorder.



Basic Chromatography

Two Important laws in chromatography

- Peak area is directly proportional to the concentration of the component.

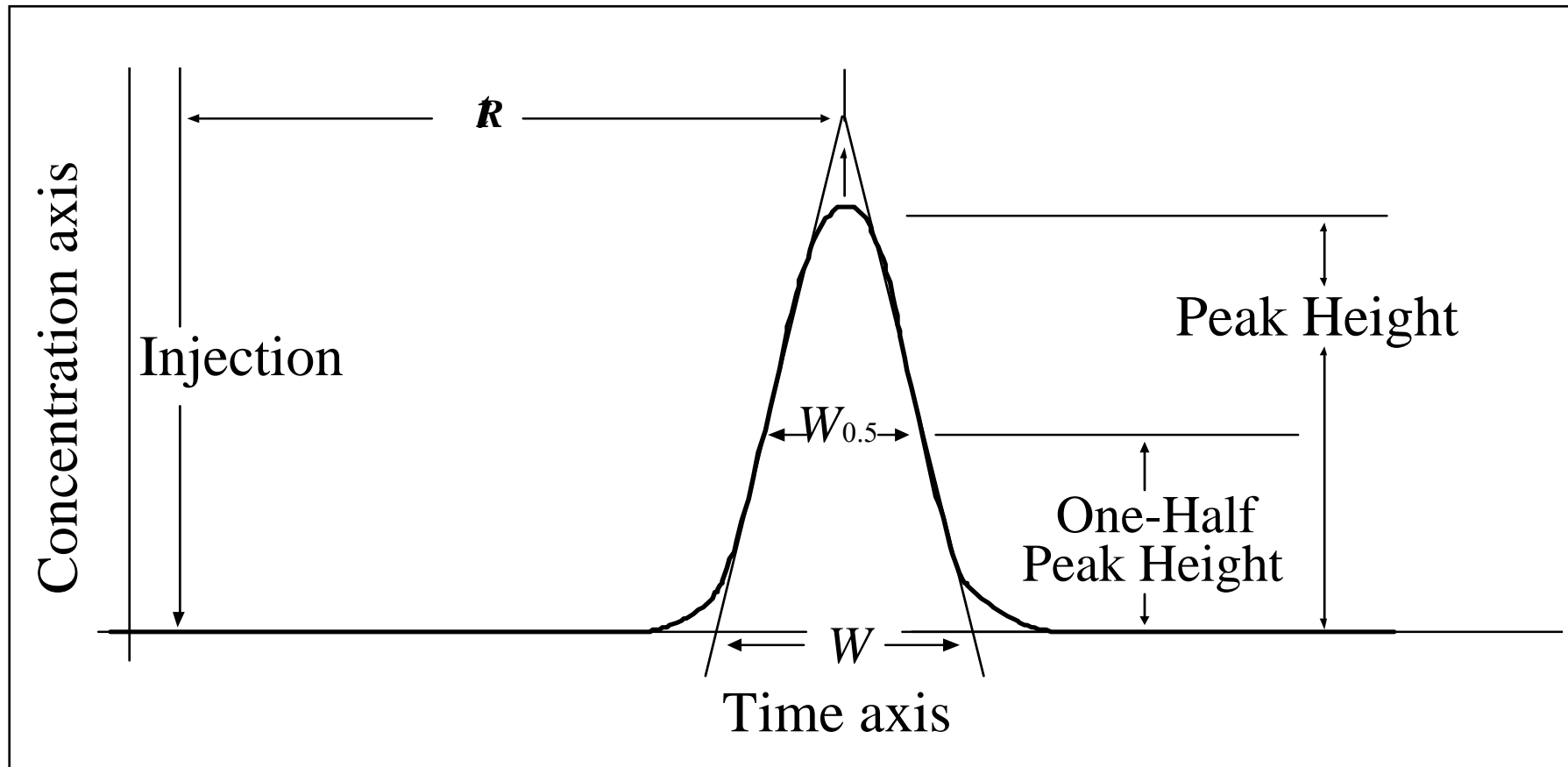
Peak Area / Response Factor = Concentration of the Component

- Retention time for each component is always the same.

Retention time is the time from sample inject to the top of the peak.

Basic Chromatography

Retention time:



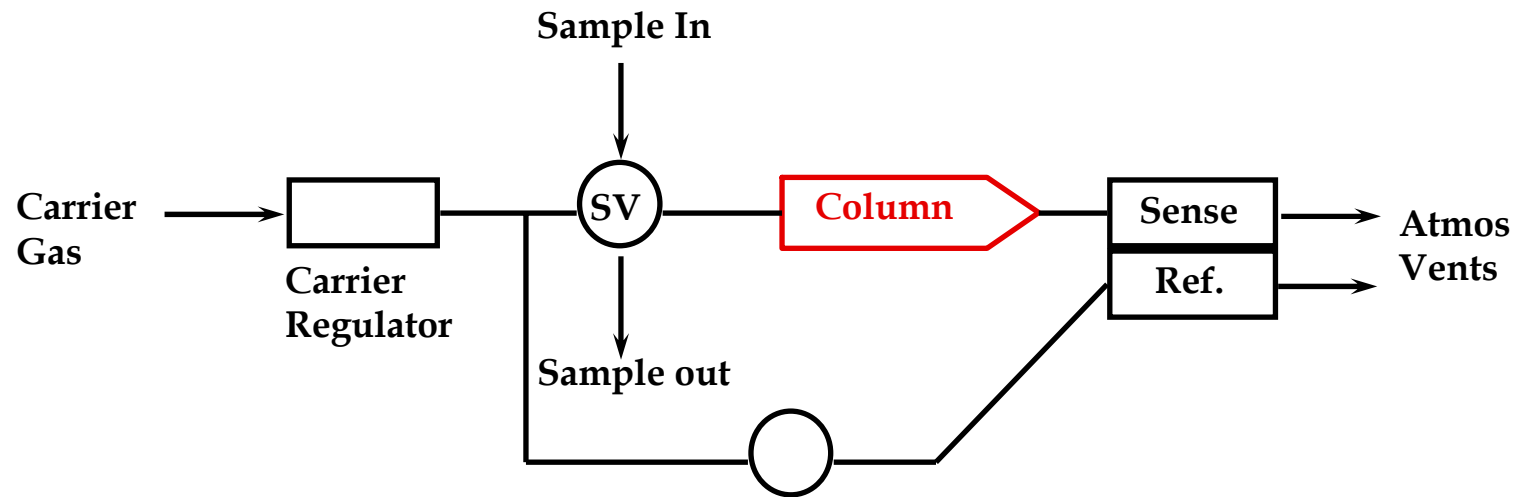
Basic Chromatography

Three things that can cause the retention time of the peak to change.

- Change in carrier flow rate.
- Change in oven temperature
- Degrading columns

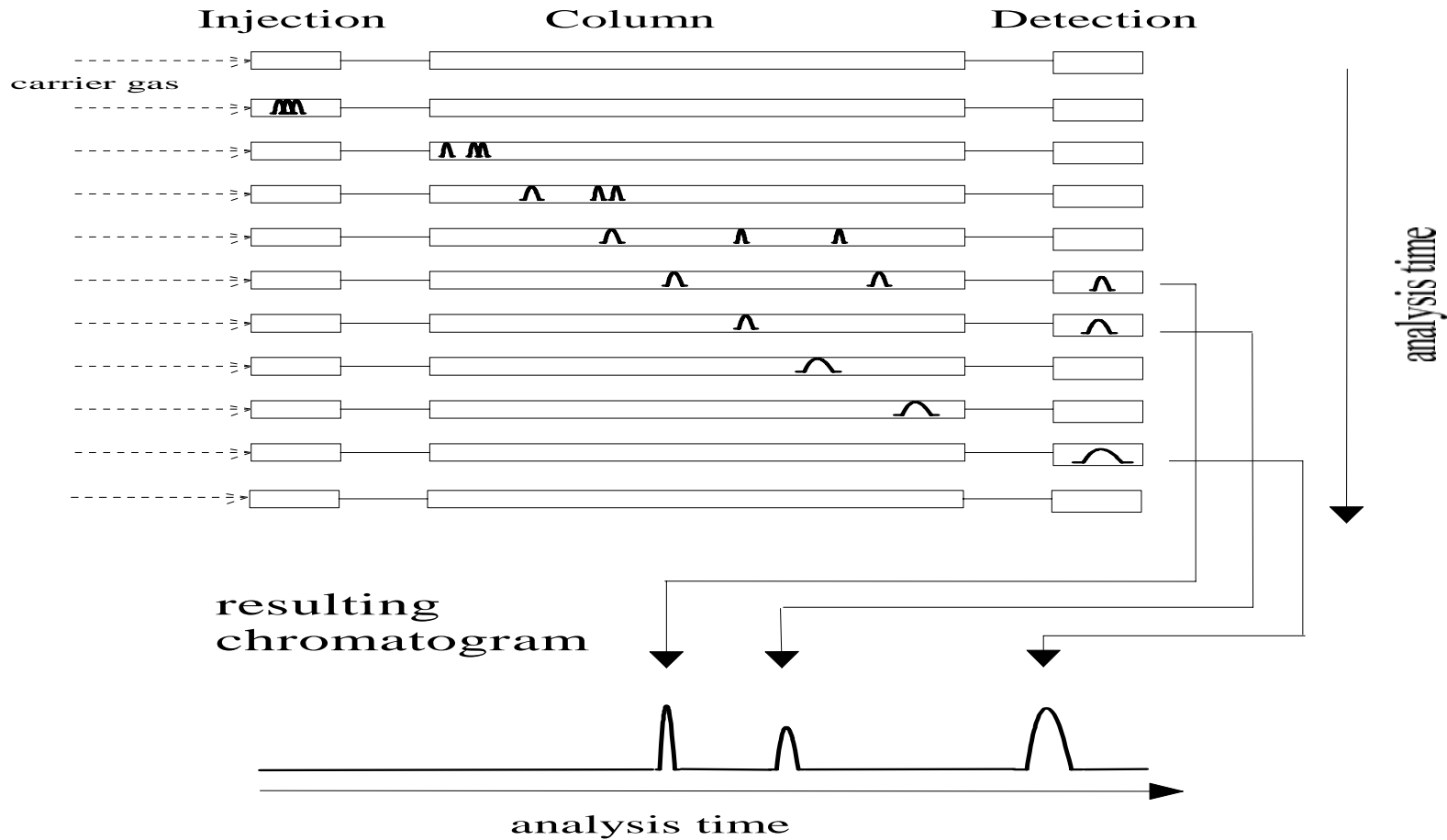
Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Column



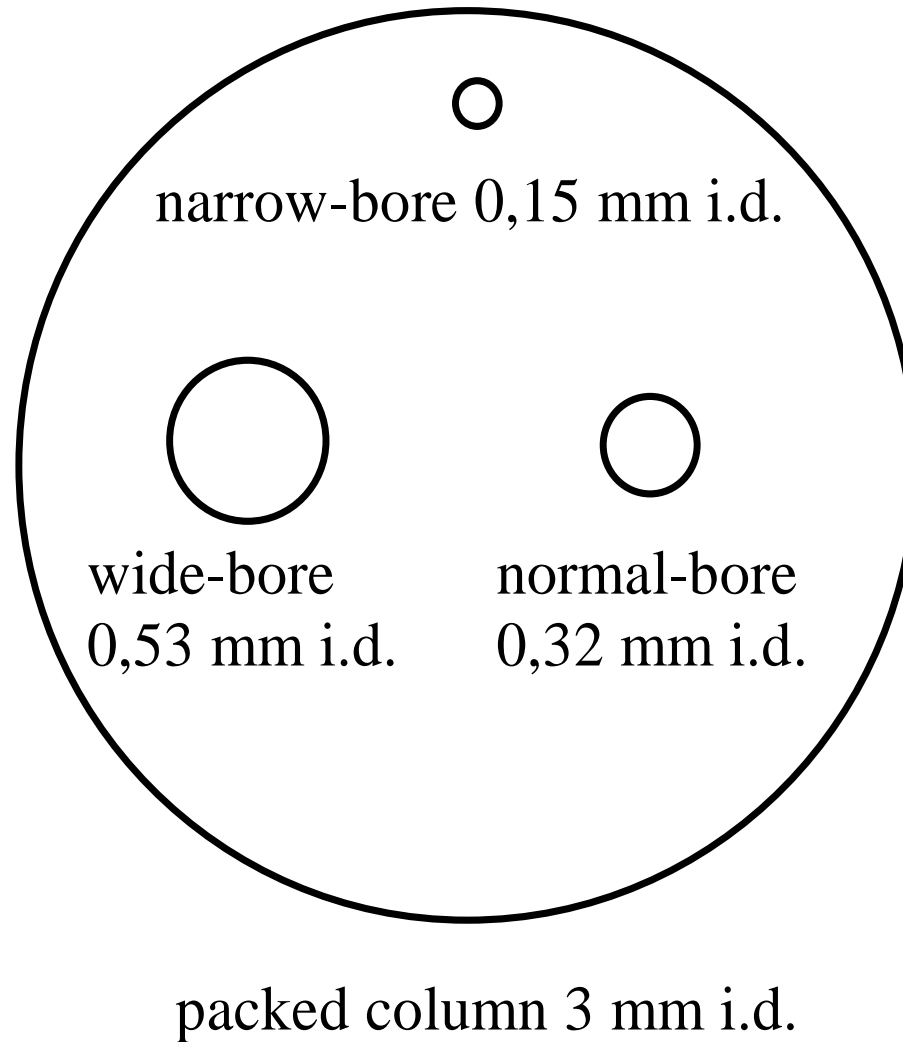
Basic Chromatography

Separation on column



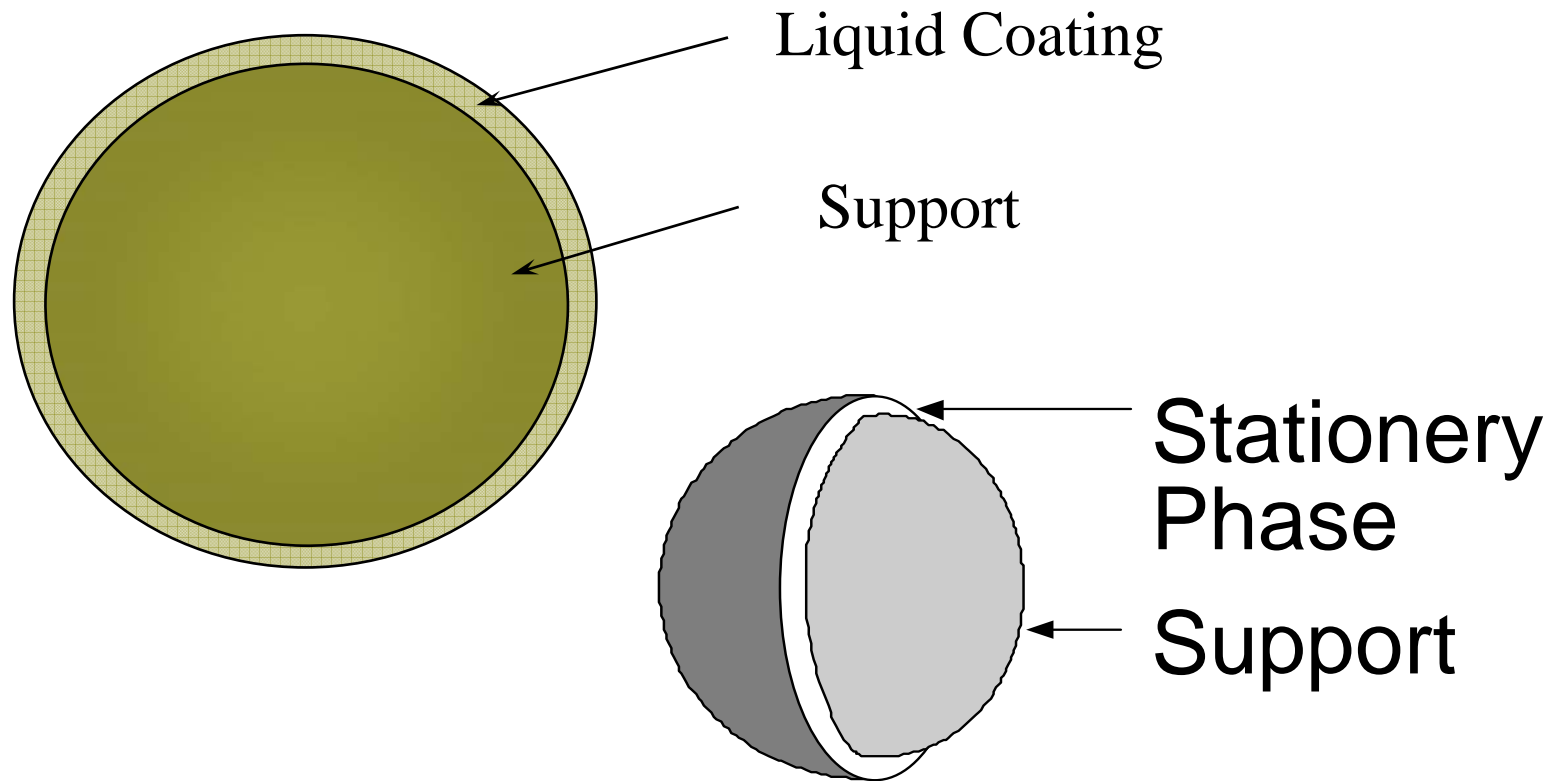
Basic Chromatography

Types of columns



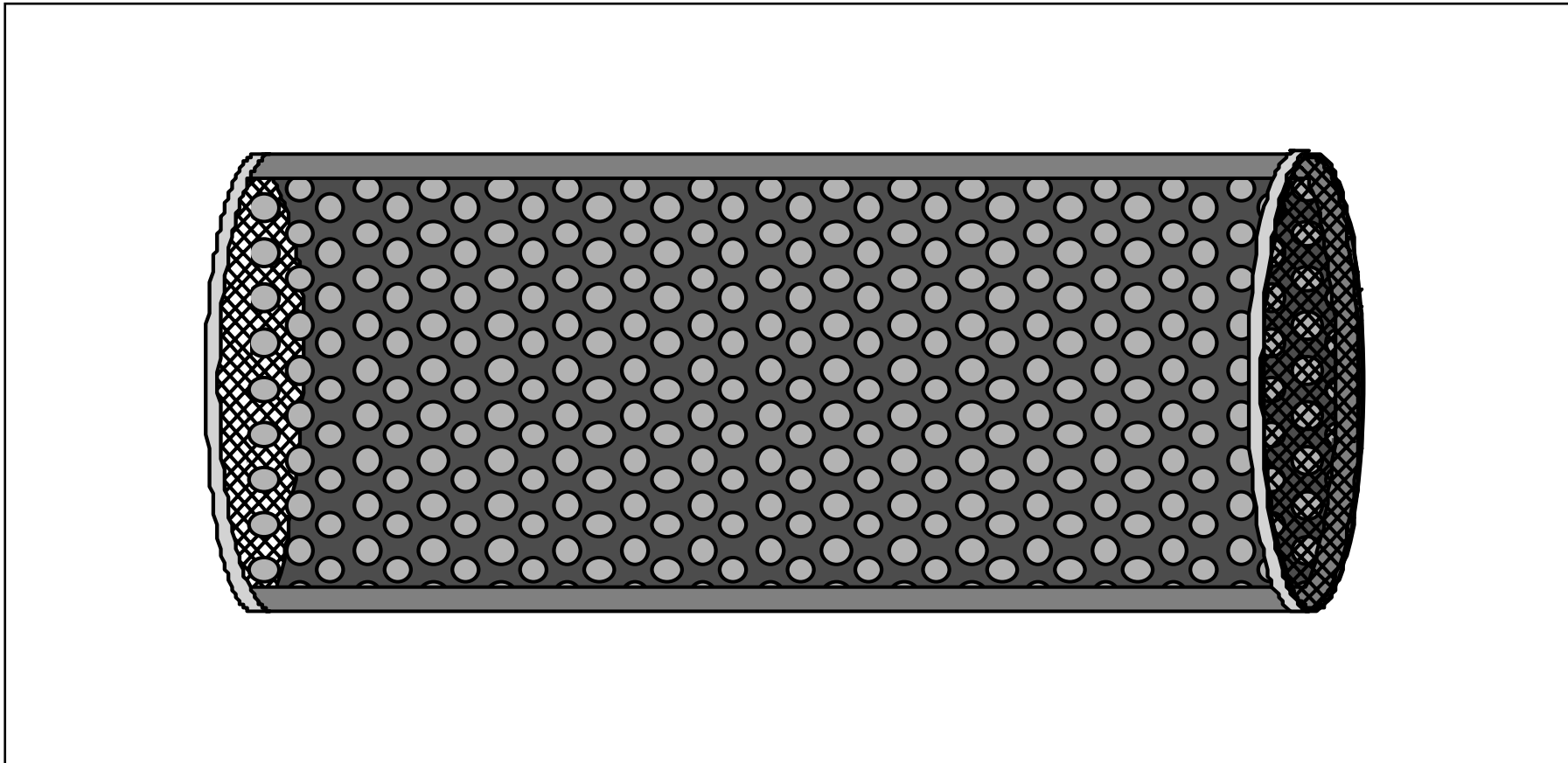
Basic Chromatography

Packed Column Material



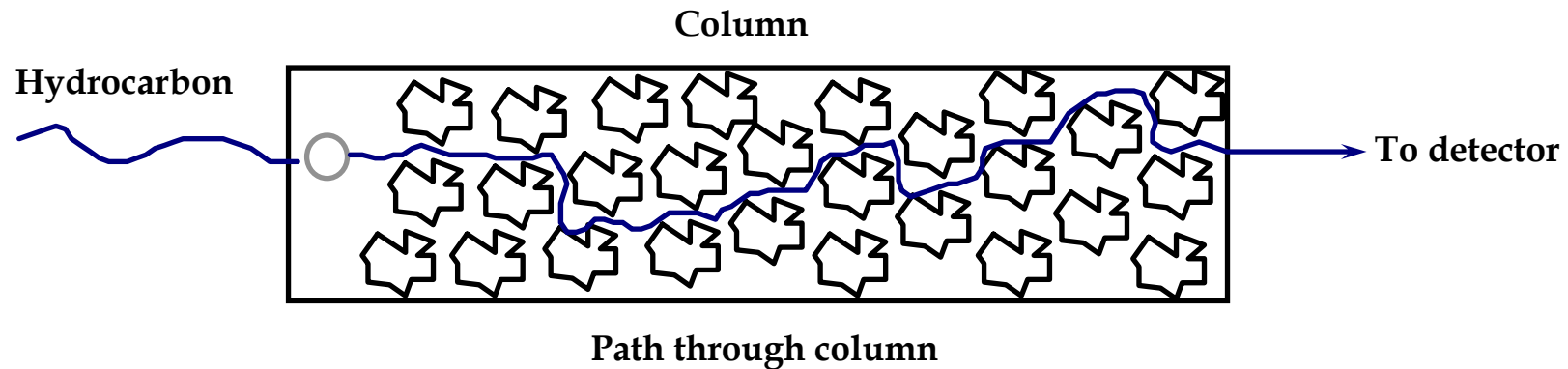
Basic Chromatography

Packed Column - Principal operation



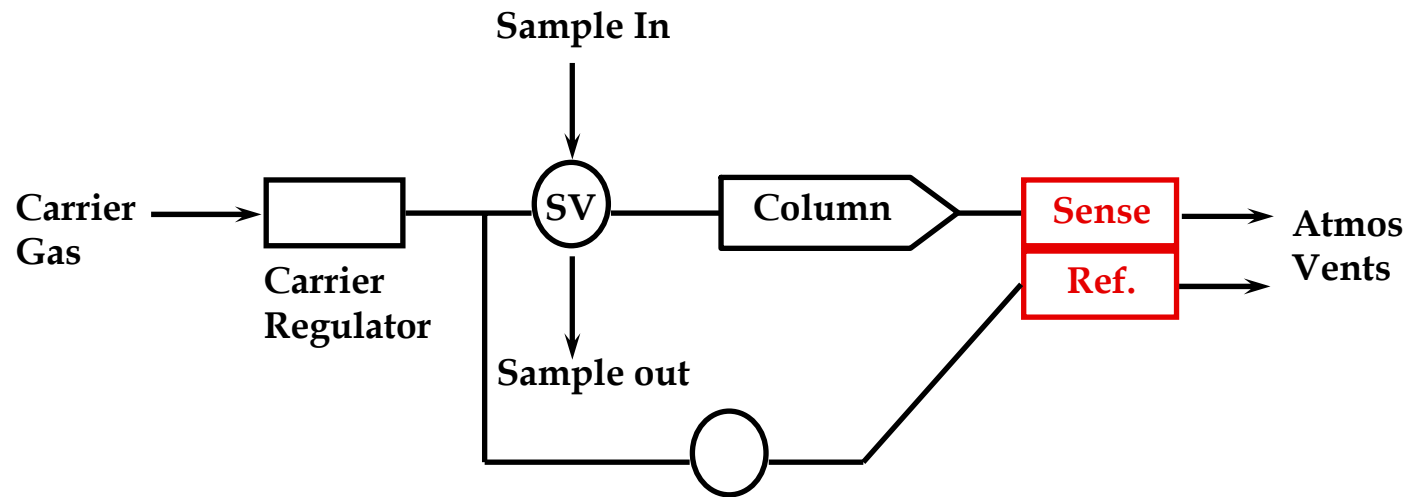
Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Column



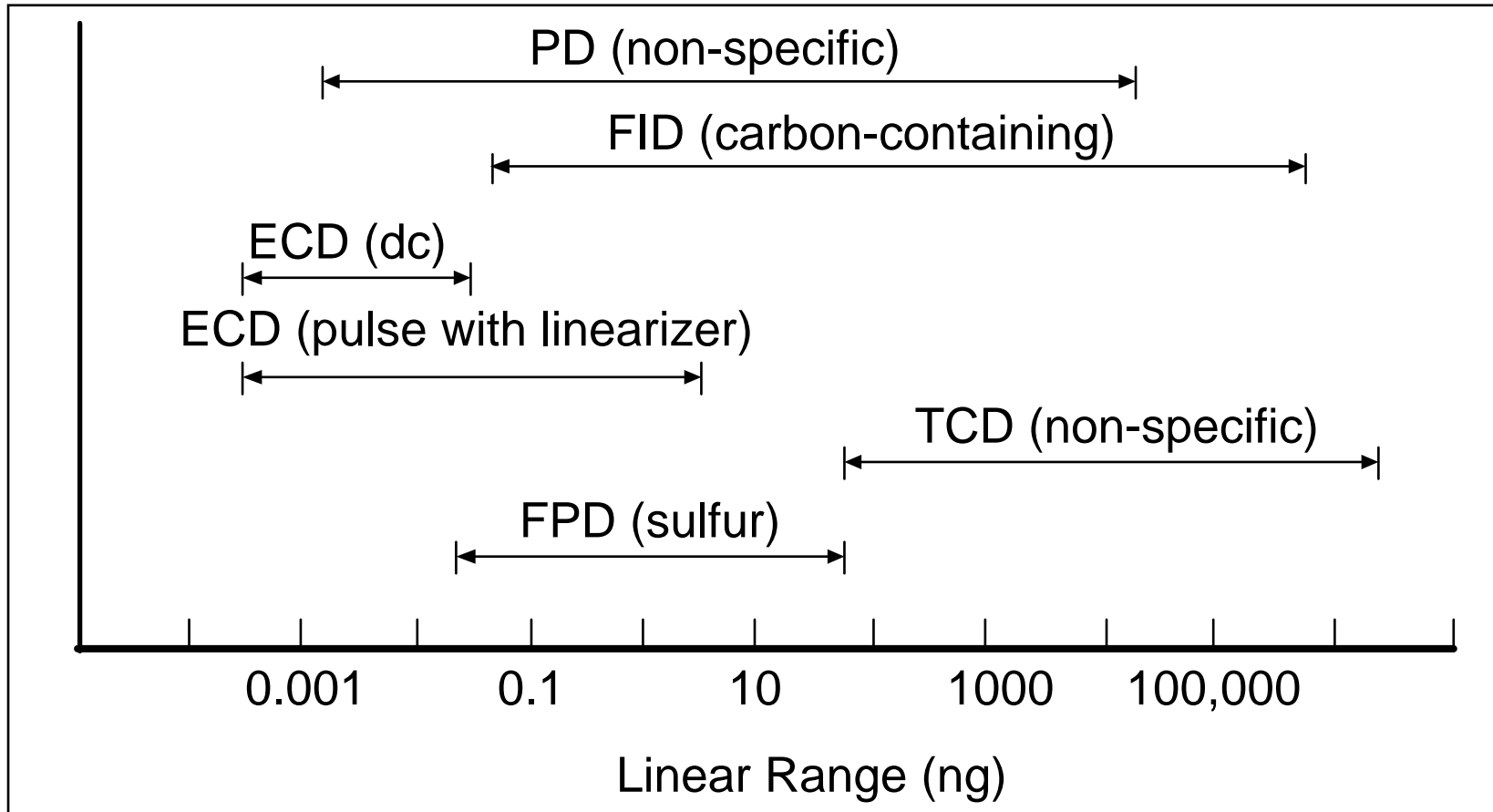
Basic Chromatography

- A basic Gas Chromatograph (GC) consists of the following parts:
 - Detector



Basic Chromatography

Detector Ranges



Basic Chromatography

Thermal conductivity: the ability of a substance to transmit heat by conduction.

■ Thermal Conductivity Detectors

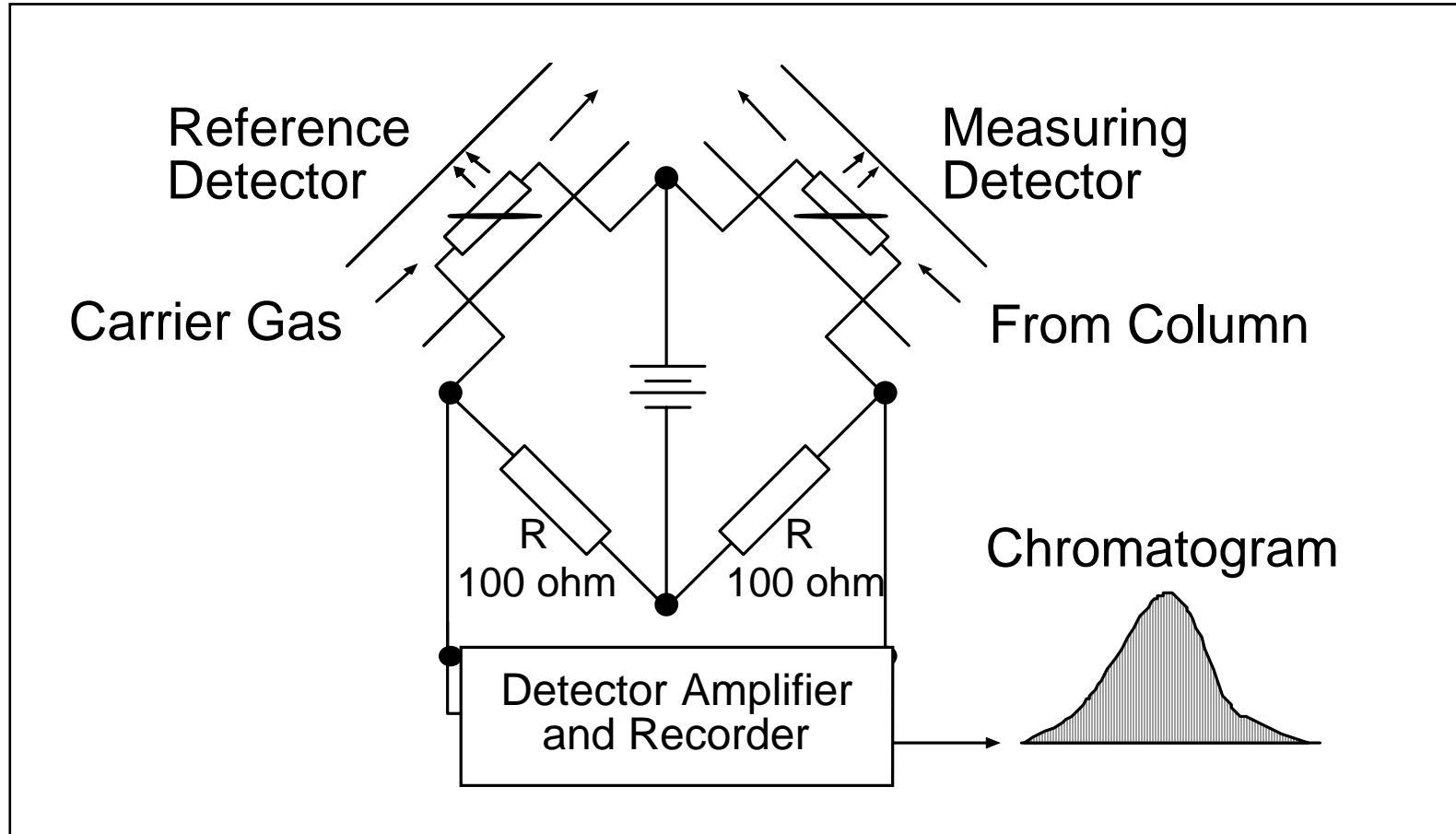
- Non-specific
- Measures the difference in thermal conductivity
- Two types
 - hot wire
 - thermistor bead

■ Advantages

- Broad applications
- low cost
- high repeatability
- low maintenance

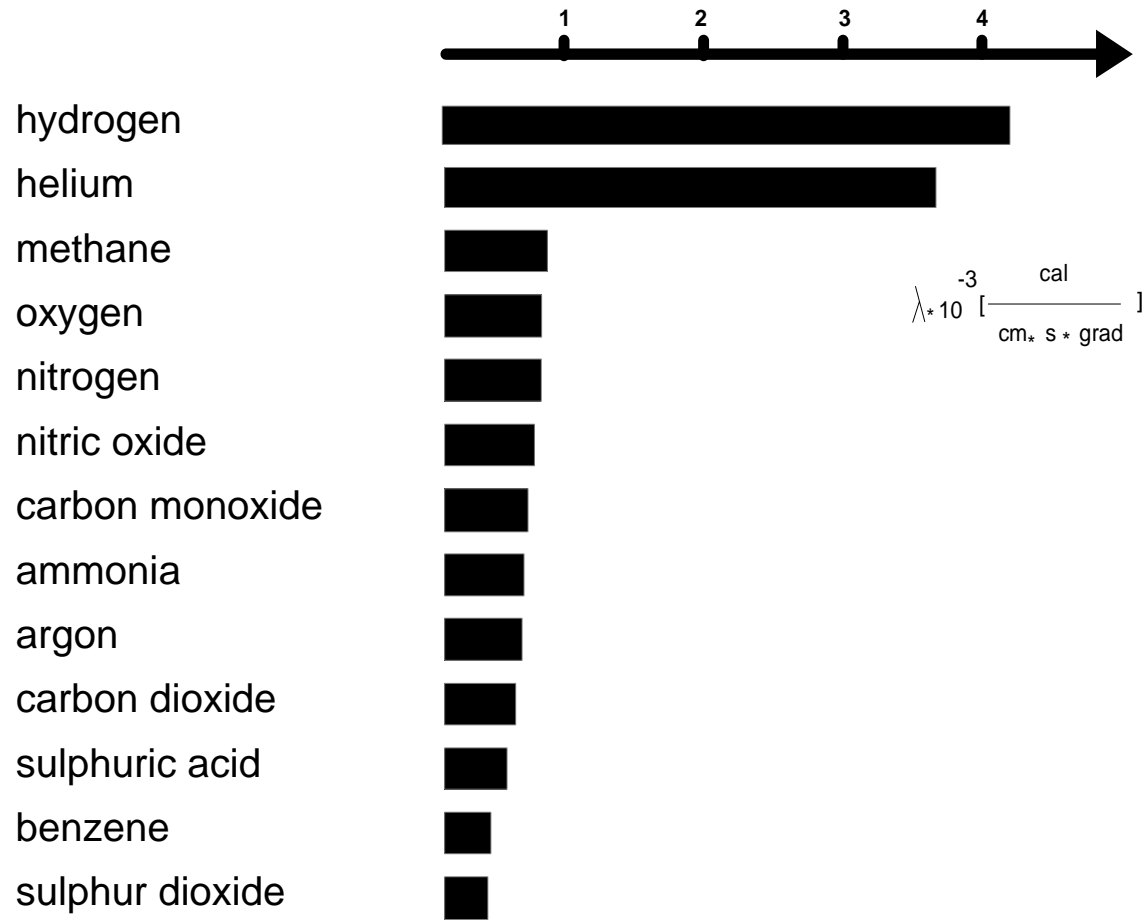
Basic Chromatography

TCD



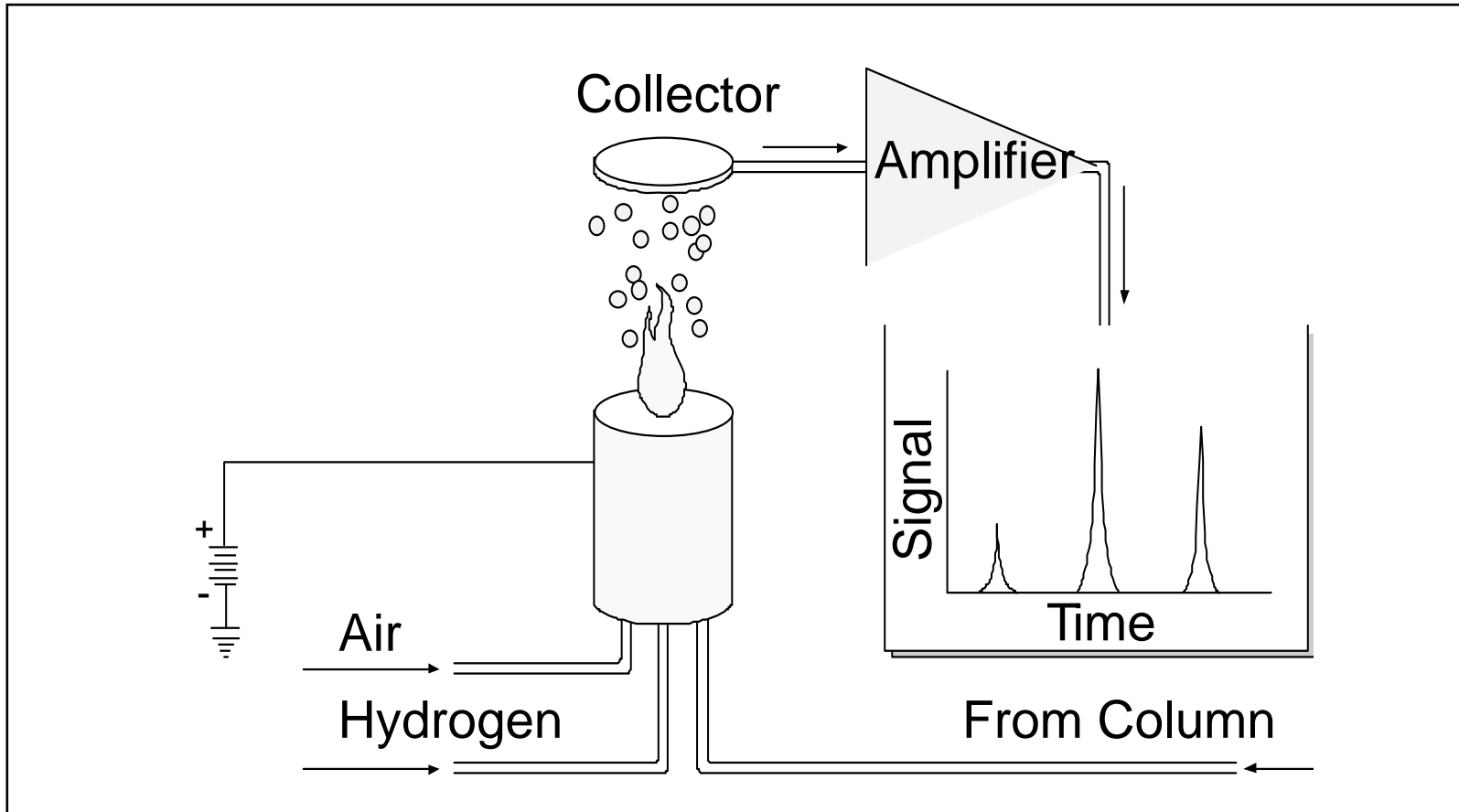
Basic Chromatography

Thermal Conductivity of some gases



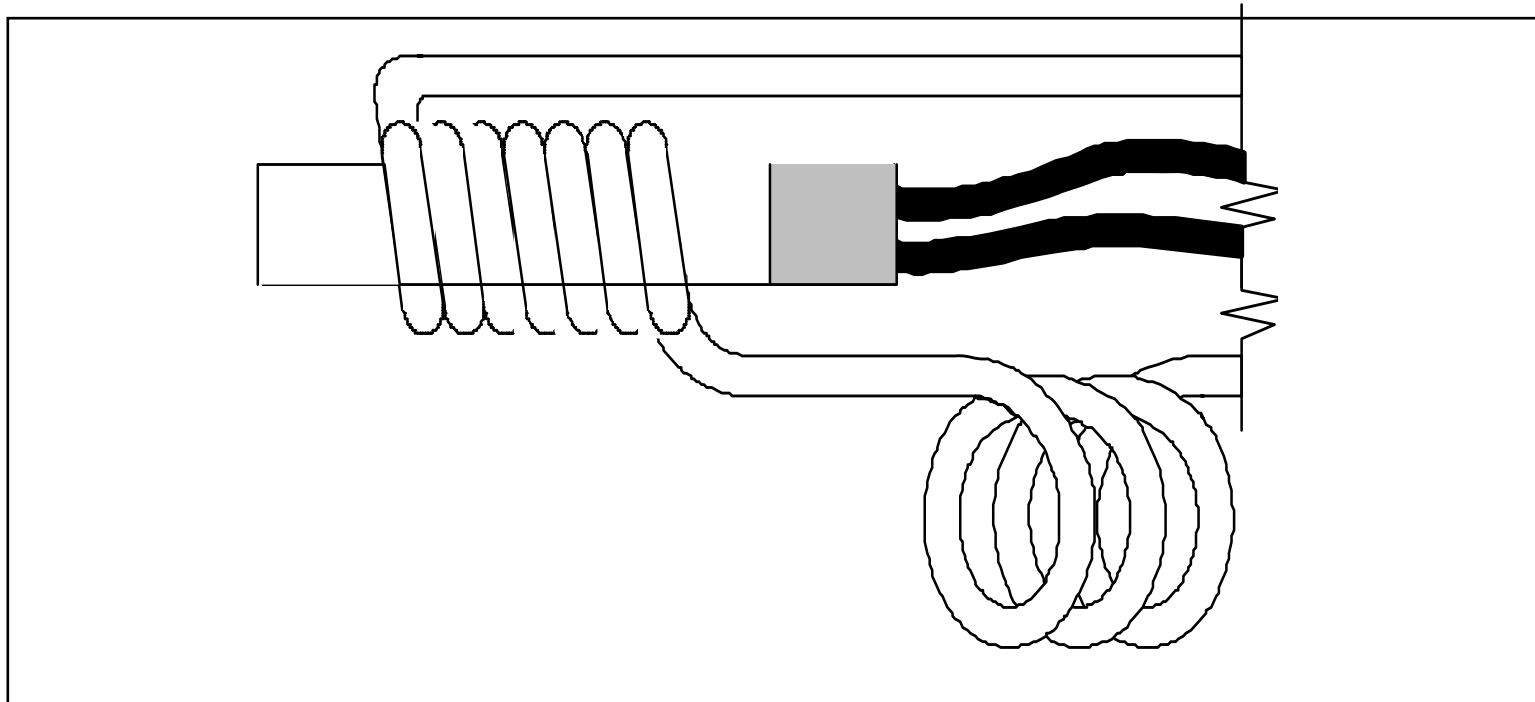
Basic Chromatography

Flame Ionisation detector



Basic Chromatography

Air Treater: Converts trace hydrocarbons in the instrument to carbon dioxide. They are converted by passing the air over a heated catalyst.



Basic Chromatography

Methanator: used with flame ionization detectors.
Converts inorganic molecules to methane by passing them through heated nickel catalyst in the presence of hydrogen.

